Subject: Chemistry

Paper name: Organic Chemistry - I

Paper No: II (CHEM/2/CC/121)

Semester: II

A. Multiple choice questions

- 1. The + I effect group among the following is
 - (a) -NO₂
 - (b) –CN
 - (c) –COCl
 - (d) –R
- 2. The most stable among the following organic intermediate carbocation is
 - (a) primary carbocation
 - (b) secondary carbocation
 - (c) tertary carbocation
 - (d) all of the above
- 3. Boiling point of intramolecular H-bonding is
 - (a) higher boiling point than intermolecular H- bonding compounds
 - (b) lower boiling point than intermolecular H- bonding compounds
 - (c) similar boiling point as intermolecular H- bonding compounds
 - (c) none of above
- 4. Reagent which have unshared pair of electron is known as
 - (a) electrophile
 - (b) nucleophile
 - (c) free radical
 - (d) none of the above
- 5. Single electron movement in organic reaction mechanism represent by
 - (a) double headed arrow
 - (b) single headed arrow
 - (c) double headed curved arrow
 - (d) single headed curved arrow

- 6. The C-C bond length in benzene is
 - (a) shorter than the C-C bond length in ethylene
 - (b) longer than the C-C bond length in ethane
 - (c) intermediate between C-C bond length in ethane and C-C bond length in ethylene
 - (d) same as that of C-C bond length in ethylene
- 7. Side chain halogenations takes place by
 - (a) electrophilic substitution mechanism
 - (b) nucleophilic substitution mechanism
 - (c) electrophilic addition mechanism
 - (d) free radical mechanism
- 8. Stability of benzene is explained by
 - (a) conjugated systems of alternating double and single bonds in benzene system
 - (b) all six p orbitals are used in the benzene π system
 - (c) reaction with KMnO₄
 - (d) all of the above
- 9. Identify the correct statement which is related to aromatic hydrocarbon
 - (a) It has only sigma bonds
 - (b) It has only pi bonds
 - (c) It has a sigma and delocalized pi bond
 - (d) none of the above
- 10. Which of the following compounds is aromatic?



(a) I and II

- (b) I and III
- (c) Only II
- (d) All
- 11. The increasing order of acidity among phenol and its derivatives are

- (a) Phenol < p-Nitrophenol < p-Cresol
- (b) p-Nitrophenol < p-Cresol < phenol
- (c) p-Nitrophenol < Phenol < p-Cresol
- (d) p-Cresol < Phenol < p-Nitrophenol
- 12. The reaction of Acetaldehyde with HCN to form cyanohydrin is an example of
 - (a) Electrophilic addition
 - (b) Nucleophilic addition
 - (c) Nuclephilic substitution
 - (d) Elimination
- 13. Among the following, the most reactive carbonyl compound towards nucleophile is:



- 14. The reactive intermediate involved in Reimer-Tiemann reation is
 - (a) Carbanion
 - (b) Carbocation
 - (c) Free radicals
 - (d) Carbene

15. Ortho-substituted benzoic acids irrespective of its electron donating or withdrawing nature are

- (a) Weaker acids than benzoic acid
- (b) Stronger acids than benzoic acid
- (c) No effect
- (d) None of these
- 16. The basicity order of the following amine is
 - (a) aniline > m-nitroaniline > p-nitroaniline > o-nitroaniline
 - (b) m-nitroaniline > p-nitroaniline > o-nitroaniline > aniline

- (c) p-nitroaniline > o-nitroaniline > m-nitroaniline > aniline
- (d) aniline > p-nitroaniline > o-nitroaniline > m-nitroaniline
- 17. Schiff's base can be obtained by reacting 1°-amine with
 - (a) Carboxylic acid
 - (b) Alcohol
 - (c) Aldehyde
 - (d) Cyanide
- 18. Secondary amine reacts with nitrous acid to form
 - (a) Alcohol
 - (b) N-Nitrosoamine
 - (c) Nitrite salt
 - (d) N, N-Dinitrosoamine
- 19. The basicity order of the following is
 - (a) $C_6H_5NH_2 < NH_3 < CH_3NH_2$
 - (b) $C_6H_5NH_2 < CH_3NH_2 < NH_3$
 - (c) $NH_3 < C_6H_5NH_2 < CH_3NH_2$
 - (d) $NH_3 < CH_3NH_2 < C_6H_5NH_2$
- 20. Carbylamine test is a confirmatory test for
 - (a) Primary amine
 - (b) Secondary amine
 - (c) Tertiary amine
 - (d) Both primary and secondary amine
- 21. S_N 2 reaction takes place with
 - (a) Strong nucleophile and 1°-alkyl halide
 - (b) Weak nucleophile and 3°-alkyl halide
 - (c) Racemisation
 - (d) Two steps
- 22. A concerted reaction is one where the
 - (a) reaction occurs in a series of steps
 - (b) reaction occurs in a single step
 - (c) reaction occurs with large threshold energy

- (d) energy of reactant is higher than the product
- 23. Which of the following is an ambident nucleophile?

(a)
$$H_2O$$

(b) \overline{NO}_2

- (d) ^{CI}
- 24. S_N1 reaction is favoured by
 - (a) polar aprotic solvent
 - (b) primary alkyl halide
 - (c) polar protic solvent
 - (d) strong nucleophile
- 25. According to Saytzeff rule, the major product in elimination reaction is
 - (a) less substituted alkene
 - (b) more substituted alkane
 - (c) less substituted alcohol
 - (d) more substituted alkene

B. Fill up the blanks

1. The movement of electrons in organic reactions can be shown

by_____notation.

- Electromeric effect is a temporary effect and it shows in organic compounds having a multiple bond in the presence______reagent only.
- 3. Hyperconjugation interaction in $(CH_3)_3C^+$ is _____ than in $CH_3 CH_2^+$.
- 4. Aryl chlorides and bromides can be prepared by electrophilic substitution of arenes with chlorine and bromine respectively in the presence of ______.
- 5. All the six carbon atoms in benzene are ______hybridized.
- 6. Hückel rule satisfy the presence of ______ p electrons in the ring and where n is an integer (n = 0, 1, 2, ...).
- 7. The hybridization state of carbonyl carbon in aldehyde and ketone
 - is_____hybridised.
- 8. Phenol is _____acidic than ethyl alcohol.

- 9. In Kolbe-schmidt reaction, phenoxide ion reacts with CO₂ to give_____.
- 10. S_N2 reaction proceeds with______of configuration.
- 11. E1 reaction is favoured by ______alkyl halide.
- 12. A reaction where more of one constitutional isomer is formed than the other is called

a_____reaction.

Key Answers

A. Multiple choice questions

- 1. (d)
- 2. (c)
- 3. (b)
- 4. (b)
- 5. (d)
- 6. (c)
- 7. (d)
- 8. (d)
- 9. (c)
- 10. (c)
- 11. (d)
- 12. (b)
- 13. (a)
- 14. (d)
- 15. (b)
- 16. (a)
- 17. (c)
- 18. (b)
- 19. (a)
- 20. (a)
- 21. (a)
- 22. (b)
- 23. (b)

24. (c)

25. (d)

B. Fill up the blanks

- 1. Curved arrow
- 2. attacking

3. more

4. FeCl₃

5. sp²

6. 4n+2

7.sp²

8. more

- 9. salicylic acid
- 10. inversion

11. tertiary

12. regioselective