Subject: Physics Paper name: Thermal and Statistical Mechanics Paper No: Phy/VI/CC/18 Semester: VI

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A. Multiple choice questions [25 (5 from each unit)]

- 1. The coefficient of viscosity of a gas varies with temperature as
  - a)  $\eta \propto T$
  - b)  $\eta \propto \sqrt{T}$
  - c)  $\eta \propto T^2$
  - d)  $\eta \propto \frac{1}{\pi}$
- 2. The most probabale velocity of a gas at NTP is 1 m/s, then its rms velocity is
  - a) 1 *m/s*
  - b) 1.225 *m/s*
  - c) 1.414 m/s
  - d) 1.5 m/s
- 3. The phenomenon of viscosity arises due to transport of
  - a) mass
  - b) momentum
  - c) energy
  - d) velocity
- 4. According to Maxwell's law of distribution of velocities, the probability of a molecule to have zero velocity is
  - a) 0
  - b) 0.25
  - c) 0.5
  - d) 1
- 5. An ideal gas possess
  - a) only kinetic energy
  - b) only potential energy
  - c) both kinetic and potential energy
  - d) neither kinetic nor potential energy
- 6. Total heat of a substance is known as
  - a) Internal energy
  - b) thermal capacity
  - c) entropy
  - d) enthalpy
- 7. At constant temperature, the  $c_p$  does not vary with pressure for
  - a) real gas
  - **b**) ideal gas
  - c) both real and ideal gas
  - d) none of these

- 8. At a given temperature, which of the following has highest thermal conductivity
  - a) hydrogen
  - b) helium
  - c) oxygen
  - d) carbon dioxide
- 9. At a given temperature, the variation of  $c_{\nu}$  with volume is zero for
  - a) real gas
  - b) ideal gas
  - c) both real and ideal gas
  - d) none of these
- 10. Which of the following is not a symbol for thermodynamic potential.
  - a) *E*
  - b) *F*
  - c) *G*
  - d) *H*
- 11. "All accessible microstates corresponding to possible macrostates are equally probable." This is the postulate of
  - a) Equal a priori probability
  - b) Equipartition of energy
  - c) Phase space
  - d) None of these
- 12. The number of accessible microstates in energy interval E and E + dE is
  - a)  $\phi(E)$
  - b)  $\sigma(E)$
  - c)  $\Omega(E)$
  - d)  $\psi(E)$

13. The six (6) dimensional space for a single particle is called

- a)  $\alpha$  –space
- b)  $\beta$  –space
- c)  $\Gamma$  –space
- d)  $\mu$  –space

14. According to statistical mechanics,  $\beta$  parameter equals

- a)  $\frac{1}{kT}$
- b) *kT*
- c)  $\frac{k}{T}$
- d)  $\frac{1}{(kT)^2}$
- 15. When systems A and B are in equilibrium, the probability of system A and B possessing energies E and E' respectively will be
  - a) zero
  - b) infinite
  - c) minimum
  - d) maximum

- 16. In Microcanonical ensemble, which of the following combination of independent variables are correct
  - a) *T* , *V* , μ
  - b) *E*, *N*, *V*
  - c) *T* , *N* , *V*
  - d)  $E, N, \Omega$
- 17. In Canonical ensemble, which of the following combination of independent variables are correct
  - a) *T*,*V*,μ
  - b) *E*, *N*, *V*
  - c) T, N, V
  - d)  $E, N, \Omega$
- 18. In Stirling's approximation, the value of  $\ln n!$  is equal to
  - a)  $n \ln n$
  - b)  $n \ln n n$
  - c)  $\ln n n$
  - d)  $\ln n n!$
- 19. The grand potential in grand canonical ensemble is given by
  - a)  $\Omega = U TS$
  - b)  $\Omega = TS$
  - c)  $\Omega = TS \mu n$
  - d)  $\Omega = U TS \mu n$
- 20. In grand canonical ensemble, for fixed p and au, Gibb's free energy is given by
  - a)  $G = \mu \tau$
  - b)  $G = \mu \rho$
  - c)  $G = \rho \tau$
  - d)  $G = \mu n$
- 21. In classical statistics, it has been assumed that all the energy levels are
  - a) accessible to all particles
  - b) accessible to selected particles
  - c) inaccessible to all particles
  - d) inaccessible to selected particles
- 22. In quantum statistics, all the energy levels are
  - a) accessible to all particles
  - b) accessible to selected particles
  - c) inaccessible to all particles
  - d) inaccessible to selected particles
- 23. The volume of a phase cell cannot be less than
  - a) *h*
  - b)  $h^2$
  - c)  $h^3$
  - d) zero
- 24. BE statistics is obeyed by
  - a) Fermions
  - b) Bosons
  - c) Baryons
  - d) Hyperons

- 25. Fermi energy at absolute zero represents the highest energy level in which all the quantum states are
  - a) unoccupied
  - b) occupied
  - c) partially occupied
  - d) None of these
- B. Fill up the blanks [15 (3 from each unit)]
- 1. The thermal conductivit ( $\kappa$ ) of a gas is \_\_\_\_\_ times the coefficient of viscosity ( $\eta$ ).
- 2. The mean free path  $(\lambda)$  of a gas varies directly as \_\_\_\_\_ and inversely as \_\_\_\_\_.
- 3. The mean free path ( $\lambda$ ) of a gas is \_\_\_\_\_ density.
- 4. Maxwell's thermodynamic relations are based on \_\_\_\_\_ thermodynamic potential(s).
- 5. Internal energy (*U*) remains constant during \_\_\_\_\_\_ processes.
- 6. Enthalpy (*H*) remains constant during \_\_\_\_\_ processes.
- In kinetic theory of an ideal gas, molecules move at \_\_\_\_\_\_ in all directions.
  At absolute zero, the molecules are in a perfect state of \_\_\_\_\_\_.
- 9. In \_\_\_\_\_, heat is transmitted from one body to another body without heating the intervening medium.
- 10. \_\_\_\_\_ law of thermodynamics states that, "if two bodies A and B are each separately in thermal equilibrium with a third body C, then A and B are also in thermal equilibrium with each other."
- 11. Amount of heat is taken to be \_\_\_\_\_, if heat is supplied to the system.
- 12. In Carnot's engine, \_\_\_\_\_\_ is the working substance.
- 13. In \_\_\_\_\_, particles are identical but distinguishable.
- 14. Fermions are particles having \_\_\_\_\_ integral spin.
- 15. In \_\_\_\_\_, particles obey Pauli's exclusion principle.

# Key Answers

A. Multiple	choice question	s [replace x]				
1. b	2. b	3. b	4. a	5. a	6. d	7. b
8. a	9. c	10. a	11. a	12. a	13. d	14. a
15. d	16. b	17. c	18. b	19. d	20. d	21. a
22. b	23. c	24. b	25. b			

- B. Fill up the blanks
- 1. specific heat at constant volume  $(c_n)$
- 2. temperature, pressure
- 3. inversely proportional to
- 4. four (4)
- 5. isochoric adiabatic
- 6. reversible isobaric adiabatic
- 7. Random
- 8. Rest
- 9. Radiation
- 10. Zeroth
- 11. Positive

12. Ideal gas
 13. MB statistics

14. half

15. FD statistics